

## Antimicrobial Potential and Metabolite Profiling in the Bioprospecting of Endophytic *Botryosphaeria rhodina* from Mangrove *Xylocarpus granatum*

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### Abstract

This research examines the antibacterial properties and metabolite diversity of *Botryosphaeria rhodina*, an endophytic fungus sourced from mangrove environments in Riau, Indonesia. The aim was to assess its antibacterial and antifungal properties and to characterize its metabolite composition for bioprospecting purposes. Solid-state fermentation was performed on a rice substrate, and crude extracts were partitioned into methanol and *n*-hexane fractions. The antimicrobial effectiveness was evaluated against eight bacterial and three fungus strains utilizing well diffusion and broth microdilution techniques. The methanol extract showed strong antibacterial activity against *Listeria monocytogenes* (inhibition zone: 28.2 mm; Minimum Inhibition Concentration (MIC): 31.25 ppm) and moderate antifungal effects against *Candida albicans*, *C. glabrata*, and *C. krusei*. The *n*-hexane extract exhibited minimal action. Metabolite profiling of the methanol extract via liquid chromatography-high resolution mass spectrometry identified over sixty compounds, including kojic acid, glycitein, fraxetin, deoxybrevianamide E, and usnic acid, whereas gas chromatography-mass spectrometry of the *n*-hexane extracts predominantly revealed fatty acids and sterols. Over twenty-five metabolites are uncharacterized, indicating the existence of unique chemical scaffolds. Partial Least Squares Discriminant Analysis established that polar metabolites were the principal contributors to antibacterial action. These findings underscore *B. rhodina* as a renewable microbial resource with considerable potential for sustainable drug discovery.

**Keywords:** Antimicrobial activity, *Botryosphaeria rhodina*, Endophytic fungi, Mangrove-derived natural products, Metabolite profiling, Solid-state fermentation, HRMS, GCMS

### Introduction

Endophytic fungi are a polyphyletic group of highly diverse ascomycetous fungi that live within plant tissues for a while without causing apparent effects. The fungi are known to be widespread among terrestrial plants, with some plants hosting hundreds of them. Endophyte-host interactions range from mutualism to parasitism depending on endophyte transmission,

infection pattern, hostage, genetic background, and environmental factors [1]. Endophytic fungi can stimulate plant growth, increase resistance to disease-causing pathogens, suppress weeds, and increase tolerance to abiotic and biotic stresses. They also can potentially produce vast bioactive secondary

metabolites with pharmaceutical importance, such as antibacterial, antifungal, and anticancer compounds [2].

Due to the unique ecosystem, endophytic fungi associated with mangrove plants are particularly interesting. Mangrove forests are a dynamic transition zone between terrestrial and marine habitats, commonly found along tropical and subtropical coastlines. Endophytic fungi from mangroves are promising sources of structurally unique natural products and drug leads with a wide range of bioactivities. The mangrove fungus may produce novel bioactive compounds like citrofulvicin, chrysogenester, brocapyrrozin A, and simpterpenoid A for antiosteoporotic, anti-inflammatory, antibacterial, and antiviral, respectively [3-6].

Mangroves are found in tropical and subtropical climates, with 41.4% of global mangroves in South and Southeast Asia. The Rhizophoraceae family includes four genera: *Bruguiera*, *Ceriops*, *Kandelia*, and *Rhizophora* [7]. Arobaya and Wanma reported Indonesia has 27% of the world's mangrove forest, or 4.25 million ha, with 143 billion ha in Riau Province [8]. The isolation and characterization of endophytic fungi from mangroves in Indonesia have been reported. More precisely, eight fungal strains were obtained from the mangrove species *Sonneratia griffithii* Kurz, located on the coast of West Sumatra. Certain ethyl acetate extracts derived from these fungi demonstrated antibacterial efficacy against *Staphylococcus aureus* and *Escherichia coli* [9]. Moreover, the antimicrobial properties of kojic acid have been isolated from the endophytic fungus *Aspergillus flavus*, which was obtained from the leaves of the mangrove plant *Sonneratia alba* on Timor Island, Indonesia. Kojic acid is widely used in the cosmetics and pharmaceutical industries due to its antibacterial properties and ability to protect against UV B radiation [10].

Our research team isolated twenty-eight fungi from two mangrove species (*Bruguiera* sp. and *Ceriops tagal*) in Sei Pakning and Tenggayun Beach, Riau Province. Among these fungi were *Trichoderma* sp. and *Penicillium* sp. Analogous assays against *S. aureus* and *E. coli* were employed to ascertain the antibacterial properties of these fungi. The findings revealed that *Trichoderma* sp. and *Penicillium* sp. demonstrated the most potent antibacterial properties [11]. In addition, *Aspergillus* sp., which has been isolated from the species

of the mangrove, showed anti-vibriosis activity against *Vibrio alginolyticus* and *V. parahaemolyticus* by using an antagonistic study [12].

For our ongoing investigation, we obtained 15 endophytic fungi from seven distinct mangrove species gathered from various locations in Riau Province. This study builds upon our prior research. The antagonistic activity of these fungi was assessed against various bacterial pathogens, such as *E. coli*, *V. parahaemolyticus*, *V. alginolyticus*, *Bacillus subtilis*, and *S. aureus*. Out of the isolated fungi, three strains displayed the highest level of effectiveness in inhibiting the growth of bacteria. By employing PCR amplification using ITS4 and ITS5 primers, we successfully identified the strains *Fusarium equiseti*, *Aspergillus fumigatus*, and *Botryosphaeria rhodina* based on the ribosomal region of the internal transcribed spacer (ITS).

Despite the growing acknowledgment of mangrove-associated endophytic fungi as significant sources of pharmacologically active secondary metabolites, the antibacterial properties and metabolite profile of *B. rhodina* remain predominantly unexamined. Prior research has predominantly concentrated on genera like *Aspergillus*, *Trichoderma*, and *Penicillium*, resulting in a paucity of extensive studies on *B. rhodina*, especially within Indonesian mangrove habitats. This knowledge deficiency constrains our comprehension of the species' metabolic capabilities and its prospective contributions to antimicrobial medication development. We propose that the secondary metabolites generated by *B. rhodina* have significant antibacterial and antifungal activity, due to a chemically varied assortment of polar and non-polar bioactive compounds. This work was conducted to assess the antibacterial properties of *B. rhodina* extracts and to execute detailed metabolite profiling utilizing liquid chromatography-tandem mass spectrometry (LC-MS/MS) and gas chromatography-mass spectrometry (GC-MS).

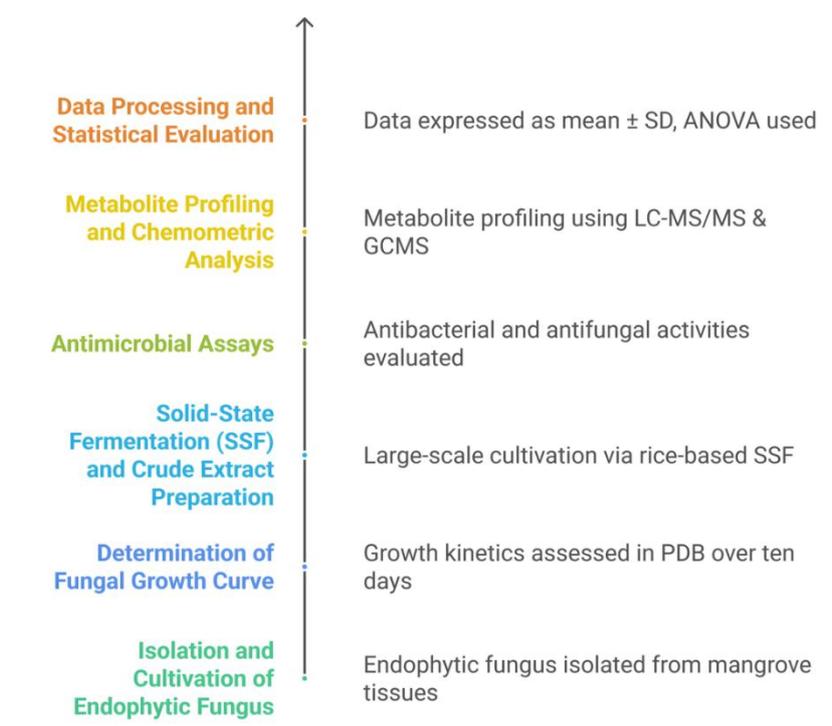
## Materials and methods

### General experimental procedure

Analytical grade solvents including methanol, *n*-hexane, ethyl acetate, and formic acid were obtained from Merck (Germany) and Sigma-Aldrich (USA). Culture media such as PDA (Potato Dextrose Agar), PDB (Potato Dextrose Broth), SDA (Sabouraud

Dextrose Agar), SDB (Sabouraud Dextrose Broth), NB (Nutrient Broth), and MHA (Mueller-Hinton Agar) were purchased from HiMedia (India). The bacterial strains used included *Staphylococcus aureus* ATCC 6538, *Bacillus subtilis* ATCC 19659, *B. cereus* ATCC 11778, *Listeria monocytogenes* ATCC 7644, *Escherichia coli* ATCC 25922, *Salmonella typhimurium* ATCC 14028, *Vibrio parahaemolyticus* ATCC 17802, and *V. alginolyticus* ATCC 17749. The fungal strains were *Candida albicans* ATCC 10231, *C. glabrata* ATCC 15126, and *C. krusei* ATCC 14243. Crude

extracts were concentrated using a rotary evaporator (Buchi Rotavapor R-300, Switzerland). Ultrasonic extraction was conducted using a Branson Ultrasonicator (USA). Metabolite profiling of methanol extract was performed using a UHPLC-Q Exactive Plus Orbitrap LC-MS/MS system (Thermo Scientific, USA), and n-hexane extract was analysed using a GCMS-QP2010S (Shimadzu, Japan). Data analysis was conducted using Compound Discoverer™ 3.3 (Thermo Fisher Scientific) and GraphPad Prism v9 (GraphPad Software, USA).



**Figure 1** Experimental workflow for the bioprospecting of endophytic botryosphaeria rhodiana propagation of endophytic fungi.

*B. rhodina* isolates were revitalized employing aseptic techniques. This study employs the direct plating technique, which entails inoculating endophytic fungus into test tubes with slanted agar media and Petri dishes to establish a stock and working culture. The PDA medium was employed. This process involves the introduction of endophytic fungus into a culture medium via the streak method and an inoculation loop. The streaked endophytic fungi were sterilized in an incubator with a 70% alcohol solution for 3 to 4 days. Certain endophytic fungi from the Petri plates and test tubes were employed for active cultivation. The

remaining fraction was maintained as a stock culture at 4 °C in a refrigerator. The endophytic fungi from the Petri dishes were transferred to test tubes employing the streak method and an inoculation loop. The test tubes were incubated for a duration of 3 to 4 days. The endophytic fungi from each test tube were transferred to a Petri dish utilizing the spread method and a spreader. The Petri dish is incubated for a duration of 3 to 4 days.

#### Growth curve of endophytic fungi

The growth curve of *B. rhodina* was determined by inoculating the fungi into test tubes containing PDB

medium. The inoculation was performed utilizing the plug method, in which the fungi were transported from Petri dishes to test tubes. The test tubes were subsequently incubated at room temperature (about 25 - 30 °C) in an incubator. In the preliminary incubation phase, a solitary endophytic fungus was isolated from each test tube and subsequently filtered by employing a vacuum filter with filter paper. Thereafter, the filtered fungus was placed in a desiccator, and their weight was recorded. The procedure was conducted daily (every 24 h) by observing the growth of the endophytic fungal isolates and measuring the changes in their dry cell weight. The measurement process ceased when the weight of the dry cell began to decrease. The weight data of the dried cells is later evaluated to ascertain the growth curve of *B. Rhodia*. The growth curve of *B. rhodina* was determined in triplicate, and the results are expressed as mean  $\pm$  standard deviation (SD). The dry cell weight was recorded daily for ten days to evaluate biomass accumulation. The mean values and SDs were plotted to construct the growth curve. Statistical validation confirmed the consistency of replicating data.

#### **Fermentation and crude extract preparation**

The fermentation of endophytic fungus was performed on solid media utilizing thirty-one 1 L Erlenmeyer flasks. Each flask contained 200 g of rice and 200 mL of sterile water. The rice in the Erlenmeyer flasks was sterilized in an autoclave at 121 °C and 1 atm pressure for 20 min to yield cooked rice. Colonies of endophytic fungi from Petri dishes were introduced into sterile water and homogenized with an inoculation loop. The suspension was thereafter inoculated uniformly onto the rice surface utilizing a micropipette. The inoculation media were incubated at ambient temperature (about 25 - 30 °C) in a sterile incubator until the stationary phase was attained. [13]

The fermented culture attains the stationary phase and is then extracted by the maceration process with ethyl acetate as the solvent. The maceration extract was concentrated using rotary evaporator at 40 °C. The crude ethyl acetate extract was subsequently partitioned via a liquid-liquid extraction technique. The crude ethyl acetate extract was solubilized in methanol. Partitioning was sustained utilizing *n*-hexane as the solvent in a 1:1 ratio until no additional components were soluble in the non-polar phase. The *n*-hexane and methanol extracts

were segregated into distinct containers, and the overall yield was determined.

#### **Antibacterial activity using well diffusion method**

The antibacterial efficacy of methanol extracts and fractions from endophytic fungus was evaluated against 4 Gram-positive pathogenic bacteria. Each pathogen was reactivated in nutritional broth and subsequently diluted to an optical density of approximately 0.1 at 600 nm. They were subsequently put to the sterile MHA medium. A cork borer was used to make 6 mm in the MHA. Subsequently, 100  $\mu$ L of extracts, a positive control (chloramphenicol at a concentration of 30 ppm), and a negative control (distilled water) were added to the MHA wells. The evaluated endophytic fungal extracts included a methanol extract at a concentration of 500 ppm and a methanol fraction at 300 ppm. The test media were then incubated at 37 °C for 12 h, with variations based on the individual pathogen being studied. The resultant zones of inhibition were further measured. The procedures were executed utilizing aseptic protocols and were replicated thrice [14,15]. All experiments were performed in triplicate ( $n = 3$ ), and the inhibition zones were expressed as mean  $\pm$  SD. The differences between treatments were statistically analyzed by one-way ANOVA and Tukey's post-hoc test ( $p < 0.05$ ).

#### **The Minimum Inhibitory Concentration (MIC) and Minimum Bactericidal Concentration (MBC)**

The antibacterial effectiveness of pure chemicals obtained from the methanol extract of the endophytic fungus *B. rhodina* was evaluated using the microdilution method. The microplate wells were augmented with extracts at a concentration of 500 ppm and a positive control of chloramphenicol at 30 ppm. Each well received 50 microliters of MHB, followed by successive dilutions to achieve compound concentrations of 50 ppm and chloramphenicol concentrations of 15 ppm. Each well was administered a total of 30 microliters of medium and 10 microliters of resazurin. Each pathogen, at a concentration of roughly 0.1 OD<sub>600</sub>, was introduced into each well in a volume of 10  $\mu$ L, excluding the media control. The samples were subsequently incubated at 37 °C for 12 h. Subsequent to incubation, 10  $\mu$ L of each chemical concentration, along with positive and negative controls, were transferred to MHA medium.

The mixture was subsequently incubated for 12 h, contingent upon the specific pathogen under examination. Bacterial proliferation was subsequently noted [14,15].

#### **Antifungal activity using well diffusion method**

The SDA was precisely dispensed into sterile petri dishes following sterilization, and the mixture was allowed to solidify. The surface of the SDA with fungi was uniformly distributed with an optical density of approximately 0.05 at 530 nm. This was accomplished with the assistance of a sterile cotton swab. The antifungal properties of the fungi-infested SDA were assessed by intentionally placing blank disc papers on its surface. The test extract was administered dropwise onto the blank discs using a micropipette containing 10  $\mu$ L of the extract solution, resulting in a final concentration of 1,000  $\mu$ g/disc. Ketoconazole was administered at a concentration of 30  $\mu$ g/mL. The negative control discs were subjected to a comparable methodology. The solutions on the discs were then permitted to cure naturally after being exposed to them for an extended period.

Incubation of the petri dishes was conducted at a consistent temperature of 37 °C for a period of 18 to 24 h following the preparation protocol. The fungal culture and the administered substances interacted efficiently during this period. Post-incubation, the petri dishes were examined for clear zones surrounding the discs, which suggested that fungal growth had been suppressed. The level of antifungal activity was precisely measured and recorded in relation to the diameters of these transparent regions. In order to ensure the reliability and consistency of the results, the test was conducted three times for each fungal strain. The samples exhibited distinct inhibition zones, which suggested the antifungal efficacy [16].

#### **Minimum Inhibitory Concentration (MIC) and Minimum Fungicidal Concentration (MFC)**

The minimum inhibitory concentration (MIC) and minimum fungicidal concentration (MFC) were determined using the microdilution technique and a 96-well microplate. At 530 nm, the optical density of *C. albicans*, *C. glabrata*, and *C. krusei* cultures was adjusted to approximately 0.05 per cent. Additionally, SDA plates were implemented to identify the MFC in

accordance with Hendra *et al.* [17] and Agustha *et al.* [18].

#### **Secondary metabolites profiling**

LC-MS/MS and GC-MS were employed to identify the secondary metabolites of the extracts. The LC-MS/MS was employed to analyze the methanol extract. The desiccated extract was subjected to ultrasonic extraction using a Branson Ultrasonic Corporation device located in Danbury, CT, USA. A mixture of 1.5 mL of methanol and approximately 50 mg of the extract was subjected to ultrasonic extraction at a temperature of 30 °C for a duration of 15 min. The resulting mixture was subsequently filtered through a 0.2  $\mu$ m syringe filter membrane produced by SY25TF PTFE mdi. Small vials were employed to capture the liquid that passed through the filter. The metabolites were analyzed using a UHPLC-Q Exactive Plus Orbitrap High-Resolution Mass Spectrometer from Thermo Scientific™ in Waltham, MA, USA, which was coupled with an Accucore™ Phenyl Hexyl (100 $\times$ 2.1 mm, 2.6  $\mu$ m) separation column. A UV detector was employed, with a wavelength of 254 nm. The ionization source for the MS analysis was the electrospray ionization method. The Q-Orbitrap mass analyzer was employed, and collision energies of 18, 35 and 53 eV were implemented. The spray voltage was maintained at 3.8 kV, the capillary temperature was maintained at approximately 320 °C, the sheath gas flow rate was 15 mL/min, the auxiliary gas flow rate was 3 mL/min, and the resolving power was 70,000 FWHM. The autosampler temperature was maintained at 10 °C, the sample injection volume was 0.5  $\mu$ L, and the flow rate of the delivery system was adjusted to 0.3 mL/min. The mobile phase was composed of acetonitrile (B) and a solution containing 0.1% formic acid in water (A). The linear gradient elution program was implemented in the following manner: The concentration of B is 5% for the initial 0 - 1.5 min. The concentration of B increases from 5% - 10% over the course of 1.5 - 9 min. The concentration of B increases from 10% - 20% between 9 and 13 min. The concentration of B increases from 20% - 28% between 13 and 17 min. The concentration of B increases from 28% - 78% between 17 and 23 min. The concentration of B decreases from 70% - 95% between 23 and 26 min. The concentration of B remains at 95% for 26 - 29 min. Ultimately, the concentration of

B decreases from 95% - 5% between 29 and 32 min. The relative abundance ranged from 0 - 100, and the exercise lasted a total of 32 min. The MS full-scan type, which operates within the mass range of 100 - 1,500 m/z, was implemented in both positive and negative ionization modes.

The Shimadzu GCMS-QP2010S instrument was used to analyze the hexane extract using gas chromatography-mass spectrometry (GC-MS). The Agilent DB-5MS UI column, which has an inner diameter of 0.25 mm and a film thickness of 0.25  $\mu\text{m}$ , was used in the analysis. The column is 30 meters in length. Helium served as the carrier gas. Electron Ionization (EI) was employed to conduct the ionization process at an energy level of 70 electron volts (eV). The gas chromatography conditions were meticulously established, beginning with an initial column oven temperature of 70.0  $^{\circ}\text{C}$  and maintaining it for 5.00 min. The temperature was gradually increased at a rate of 5.00  $^{\circ}\text{C}$  per minute until it attained a final temperature of 305.0  $^{\circ}\text{C}$ . Subsequently, the temperature was maintained at this level for precisely 18 min. The injection parameters consisted of a split injection mode at a temperature of 300.0  $^{\circ}\text{C}$  and a split ratio of 49.0. The flow control was effectively maintained in pressure mode, with a total flow rate of 35.6 mL/min and a column flow rate of 0.65 mL/min. The linear velocity was 29.6 cm/s as a result of the pressure adjustment to 30.0 kilopascals. The discharge rate of the purge was established at 3.0 mL/min. In order to ensure the system's readiness, the column oven, SPL1, and MS units were examined, and the SPL1 carrier and purge flow were subjected to supplementary inspections.

For mass spectrometry, the interface temperature was maintained at 305.0  $^{\circ}\text{C}$ , while the ion source temperature was maintained at 250.0  $^{\circ}\text{C}$ . The acquisition parameters were set to commence at 3.20 min and conclude at 70.00 min in scan mode, with a scan speed of 1,250 and an event time of 0.50 s. The scanning mass range was established as 28.00 to 600.00 m/z. The solvent cutting process was scheduled to last for 3 min. The detector was operated in absolute gain mode, with a detector gain of 1.50 kV and a threshold of 0. In order to ensure the precision and stability of the measurements, the system implemented a 3.0-minute equilibrium period.

### Data analysis.

The mean value plus or minus the standard deviation (SD) was used to convey the antimicrobial activity results obtained using the well diffusion method. GraphPad Prism Version 9 was employed to analyse the data. The means were contrasted using a one-way analysis of variance (one-way ANOVA), followed by Tukey's test. Statistically significant values were defined as those with a  $p < 0.05$ . Compound Discoverer 3.3 software (Thermo Fisher Scientific) was employed to analyze mass spectrometry data. At a mass tolerance of 5 ppm, features were identified and categorized by retention duration, with a tolerance of 0.2 min. The ChemSpider database was employed to identify the compounds. The mzCloud database was used to correlate MS2 spectra in order to improve chemical annotation. A multivariate chemometric study was conducted utilizing the online MetaboAnalyst 5.0 platform ([www.metaboanalyst.ca](http://www.metaboanalyst.ca)). Partial Least Squares Discriminant Analysis (PLS-DA) was utilized to distinguish the metabolite profiles of methanol and n-hexane extracts using LC-HRMS and GC-MS datasets. Variable Importance in Projection (VIP) scores were computed to ascertain the discriminant metabolites that most significantly contribute to class differentiation. Score plots and heatmaps were created to illustrate clustering tendencies in bacterial and fungal bioassays.

## Results and discussion

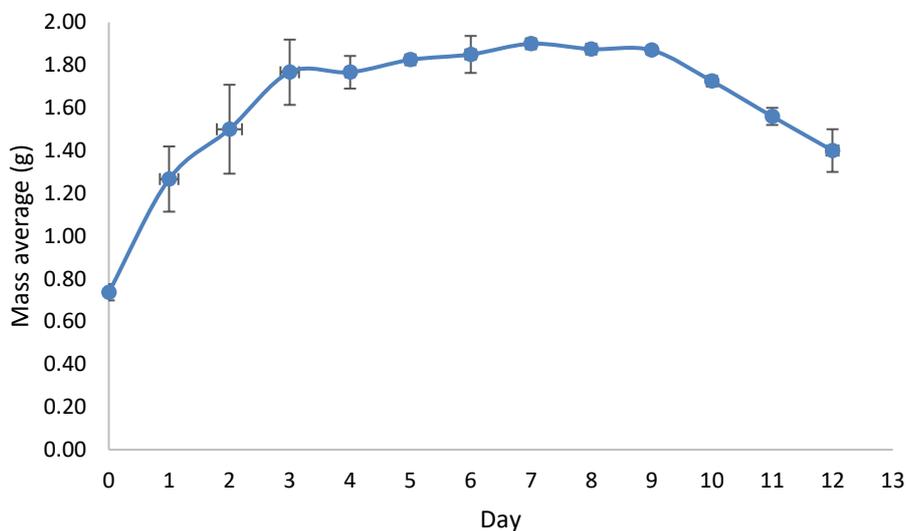
### Growth curve of endophytic fungi

Understanding fungal growth kinetics is crucial to optimize secondary metabolite production. This work constructed the growth curve of *B. rhodina*, isolated from mangrove plants in Riau Province, to identify the ideal harvesting time for maximizing metabolite yield. The fungus was grown in Potato Dextrose Broth (PDB) and kept at ambient temperature (about 25 - 30  $^{\circ}\text{C}$ ). Biomass accumulation was evaluated by daily measuring the dry cell weight during a ten-day duration.

The growth curve exhibited a typical sigmoidal pattern, with distinct lag, exponential, stationary, and decline phases, based on triplicate measurements of dry biomass (mean  $\pm$  SD). The stationary phase (days 6 - 8) was identified as the optimal window for harvesting biomass for subsequent solid-state fermentation, where secondary metabolite biosynthesis was maximal.

As seen in **Figure 2** the growth curve of *B. rhodina* showed a typical sigmoidal pattern with four separate phases: lag, exponential (log), stationary, and decline. During the lag phase (days 0 - 2), biomass increased

slightly as the fungus adapted to the nutrient-rich medium. This phase is marked by vigorous metabolic reprogramming and enzyme production, with limited cell division.



**Figure 2** Growth curve of *B. rhodian*.

The exponential phase, occurring between days 3 and 6, exhibited a significant rise in dry biomass, signifying rapid hyphal growth and elevated metabolic activity. This phase signifies ideal circumstances for primary metabolism, encompassing the production of vital macromolecules necessary for cellular growth and division. The curve's slope during this phase indicates the organism's maximal specific growth rate ( $\mu_{max}$ ) under the specified culture conditions.

The curve stabilized between days 6 and 8, indicating the start of the stationary phase. This phase is very crucial for the formation of secondary metabolites. In nutrient-deficient situations and heightened metabolic stress, fungi frequently activate secondary metabolism as a survival mechanism. In *B. rhodina*, this phase is posited to align with the production of antimicrobial chemicals, subsequently validated by LC-MS/MS and GC-MS analysis of the extracts. The stationary phase represents a crucial opportunity for the collection of fungal biomass for metabolite extraction, as the expression of biosynthetic gene clusters often reaches its zenith during this phase [19].

Following the stationary phase, the dry biomass declined after day 9, indicating the onset of the dying

phase. This transition reflects metabolic exhaustion and cellular degradation typical of late growth stages. In this phase, autolysis and the breakdown of cellular components prevail, resulting in a decrease in biomass. Harvesting at this stage is inadequate, as both cell viability and metabolite integrity may be jeopardized [20].

The combining of the growth curve with the fermentation approach is essential. Fermentation was carried out on solid-state media (rice substrate), initiated during the stationary phase—predicted from the PDB growth curve—to optimize secondary metabolite yield. Such an approach aligns with previous reports linking the onset of the stationary phase to enhanced production of polyketides, alkaloids, and terpenoids in filamentous fungi [21].

The growth curve of *B. rhodina* is an essential instrument for determining the optimal period for biomass harvest and the extraction of bioactive secondary metabolites. **Figure 2** distinctly demonstrates that the period from day 6 to day 8 constitutes the ideal harvest window, coinciding with the stationary phase, during which the maximum potential for secondary metabolite accumulation is anticipated. Establishing

these growth dynamics improves repeatability, aids subsequent bioactivity testing, and guides future scale-up methods for the medicinal utilization of mangrove-derived endophytic fungus.

#### **Solid state fermentation process of fungi**

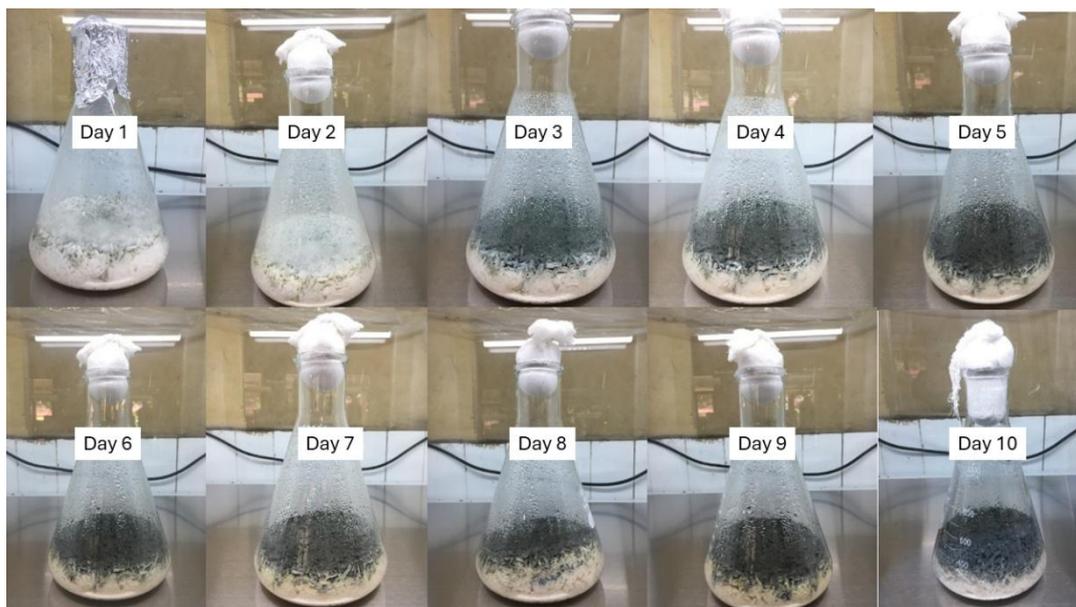
Following the determination of *B. rhodina*'s growth curve in liquid culture, which revealed that the stationary phase occurred between days 6 and 8 (**Figure 2**), the next step was to implement solid-state fermentation (SSF) to optimize the production of bioactive secondary metabolites. Compared with submerged fermentation, solid-state fermentation offers higher yields, lower contamination risk, and reduced cost. Furthermore, SSF more closely resembles the normal ecological niche of endophytic fungi, particularly those found in plant tissues, which frequently thrive on solid organic substrates.

Rice was selected as the substrate for solid-state fermentation due to its favorable physicochemical characteristics that support the growth and metabolic activity of filamentous fungi. Its balanced carbon-to-nitrogen ratio, high carbohydrate content, and porous structure provide an extensive surface area for fungal colonization and efficient oxygen transfer. The moderate moisture in rice mimics the natural habitat of endophytic fungi and promotes biosynthetic gene cluster activation. Previous studies have demonstrated that rice-based substrates enhance metabolite diversity and yield in fungal fermentations, particularly for the biosynthesis of polyketides, terpenoids, and alkaloids [21,22]. A total of about 40 Erlenmeyer flasks (1 L capacity) were filled with 200 g of sterilized rice and 210 mL of distilled water. The inoculum of *B. rhodiana*, previously cultured

on PDA, was equally distributed across the rice surface. The flasks were incubated at room temperature (25 - 30 °C) for ten days. During the incubation period, apparent mycelial development gradually colonized the rice matrix, transforming it from white to a densely coated greyish-black fungal mass (**Figure 3**).

All solid-state fermentations were carried out under controlled conditions to ensure reproducibility. The initial moisture content of the rice substrate was adjusted to  $60 \pm 2$  % (w/w) and maintained by periodic humidification to prevent drying during incubation. The inoculum density was standardized to 5 % (v/w) using an actively growing mycelial suspension, ensuring uniform colonization across all flasks. Incubation was performed at  $28 \pm 2$  °C under ambient aeration without forced airflow, reflecting optimal conditions for *B. rhodina* growth and metabolite synthesis. An uninoculated rice substrate (control) was processed in parallel following identical extraction procedures to exclude potential interference from the substrate matrix. Extraction yields were normalized to the dry weight of the fermented substrate (mg extract/g dry biomass) to enable quantitative comparison between replicates.

Over a 10-Day Incubation Period displays the morphological evolution of fungal biomass during the fermentation process. On day two, fungal colonization began, and by day four, there was significant surface covering. Maximum mycelial density and sporulation occurred between days 6 and 8, which corresponded to the previously observed stationary period under submerged conditions. This lends support to the hypothesis that harvest timing is an important component in synchronizing maximal biomass output with secondary metabolite accumulation.



**Figure 3** Solid-State fermentation of *B. rhodina* on rice substrate over a 10-Day incubation period.

The application of solid-state fermentation (SSF) for the synthesis of secondary metabolites is well documented in the literature. The reduced moisture levels in solid-state fermentation systems promote the synthesis of metabolites such as polyketides, terpenoids, and alkaloids, which are frequently linked to antibacterial and cytotoxic properties [21,23]. The restricted diffusion of oxygen and food gradients in solid substrates may act as physiological stimuli for the activation of dormant biosynthetic gene clusters, thus increasing chemical diversity. [24].

Following fermentation, the fungal biomass, along with the colonized rice substrate, underwent extraction by the maceration process with ethyl acetate (EtOAc) as the solvent. This solvent was chosen for its moderate polarity and effectiveness in extracting a wide array of semi-polar secondary metabolites [25]. The maceration process was conducted in three consecutive cycles, each lasting 24 h. The initial extraction utilized 1,000 mL of EtOAc, succeeded by 800 mL and 600 mL in the subsequent second and third extractions, respectively. The amalgamated filtrates were subsequently concentrated under diminished pressure utilizing a rotary evaporator at 40 °C, resulting in a semi-solid crude ethyl acetate extract.

The whole crude extract was then weighed and allocated for additional processing. Approximately 1 g of the crude extract was designated for qualitative screening of secondary metabolites, indicating the

presence of terpenoids, phenolic compounds, and flavonoids. The remaining extract underwent liquid-liquid partitioning to isolate components according to their polarity. The crude extract was solubilized in methanol, and partitioning was performed using *n*-hexane in a 1:1 ratio. The operation was reiterated until no additional components were transported to the non-polar phase, signifying complete extraction.

The partitioning method produced two parts with differing physicochemical characteristics. The *n*-hexane fraction, indicative of the non-polar component of the extract, had a yellowish-brown hue and released a strong odor. The final yield of the hexane fraction after evaporation was 37.17 g. This fraction probably includes lipophilic chemicals such as fatty acids, terpenoids, and low polarity phenolics. These families of chemicals have been previously documented to demonstrate antibacterial, antioxidant, and anti-inflammatory effects. [26].

Conversely, the methanolic fraction, which maintained the more polar elements, displayed a dark brown to black hue and emitted a faintly fragrant odor. Upon solvent evaporation, the ultimate yield of this fraction was noted to be 56.19 g. The methanol fraction is believed to encompass polar bioactive molecules, including polyphenols, flavonoids, alkaloids, and polyketides, which are well-documented for their significant antibacterial and bioactive characteristics. [27,28]. The superior yield of the methanol fraction

relative to the n-hexane extract likely signifies the prevalence of polar and semi-polar secondary metabolites in *B. rhodina*, highlighting the fungus's potential as a prolific source of various natural products for pharmacological investigation.

Several critical findings arose from the solid-state fermentation (SSF) and subsequent extraction procedures. The identification of the ideal biomass harvest date, informed by the previously established growth curve in liquid culture (**Figure 2**), greatly enhanced secondary metabolite production. The collection of fungal biomasses during the stationary phase (days 8 - 10) coincided with the maximum production of secondary metabolites, as demonstrated by the significant quantity of crude extracts acquired. The sequential maceration and solvent-partitioning approach enabled efficient recovery and fractionation of a wide array of metabolites. Ethyl acetate served as an effective initial solvent, encompassing a broad range of metabolites, while further partitioning with methanol and n-hexane enhanced separation according to chemical polarity.

The unique physicochemical properties, such as color, flavor, and yield of the resultant fractions, signify varying secondary metabolite profiles, which are essential for future bioassay-guided separation and structural elucidation. The integration of these fractions

with sophisticated analytical platforms like liquid chromatography-tandem mass spectrometry (LC-MS/MS) and gas chromatography-mass spectrometry (GC-MS) is crucial for thorough metabolomic characterization. The effective execution of rice-based SSF for *B. rhodina* highlights the potential for scaling. This fermentation technique, noted for its simplicity, cost efficiency, and scalability, offers a promising strategy for future commercial applications in the pharmaceutical, agricultural, and biotechnology industries.

### Antimicrobial activities

Antibacterial activity of *B. rhodina* extracts was tested against eight clinically relevant bacteria using well diffusion and broth microdilution methods. **Table 1** demonstrates that the methanol extract displayed a wider and more potent antibacterial efficacy than the n-hexane extract. In the case of *Listeria monocytogenes*, the inhibition zone measured  $28.20 \pm 0.40$  mm, signifying robust activity ( $\geq 16$  mm), whereas the n-hexane extract exhibited negligible inhibition ( $6.21 \pm 0.4$  mm, classified as inactive). A like tendency was observed for *Bacillus subtilis*, *Staphylococcus aureus*, and *Vibrio parahaemolyticus*, wherein the methanol extract consistently yielded moderate to high inhibitory zones, measuring between 16.47 and 21.37 mm.

**Table 1** Antibacterial activities of various extracts from *B. rhodina* at concentrations 1,000 ppm.

No.	Microorganism	Clear zone diameter (mm)		
		n-Hexane extract	Methanol extract	Chloramphenicol 30 ppm
1	<i>B. cereus</i>	$13.21 \pm 0.5^c$	$18.86 \pm 0.3^b$	$26.35 \pm 2.1^a$
2	<i>B. subtilis</i>	$17.62 \pm 1.1^c$	$21.37 \pm 0.81^b$	$31.97 \pm 2.1^a$
3	<i>E. coli</i>	$6.72 \pm 0.31^c$	$9.91 \pm 0.64^b$	$28.6 \pm 0.4^a$
4	<i>L. monocytogenes</i>	$6.21 \pm 0.4^c$	$28.2 \pm 0.4^b$	$31.2 \pm 2.1^a$
5	<i>S. typhimurium</i>	$13.25 \pm 0.4^c$	$16.47 \pm 0.09^b$	$22.8 \pm 1.1^a$
6	<i>S. aureus</i>	$16.08 \pm 0.6^c$	$18.41 \pm 0.4^b$	$28.1 \pm 3.1^a$
7	<i>V. alginolyticus</i>	$6.25 \pm 0.4^c$	$19.18 \pm 3.2^b$	$21.95 \pm 3.1^a$
8	<i>V. parahaemolyticus</i>	$14.36 \pm 0.3^c$	$17.86 \pm 1.0^b$	$25.82 \pm 2.3^a$

The results were expressed as mean  $\pm$  Standard Deviation (SD) in the same row with different superscript letters indicates a statistically significant difference based on Tukey's test ( $\alpha = 5\%$ ).

The *n*-hexane extract shown moderate antibacterial activity solely against *Bacillus subtilis* (17.62 mm) and *S. aureus* (16.08 mm), while its effectiveness against other examined bacterial strains varied from inactive to weak, with inhibition zones of  $\leq$  15 mm. Conversely, chloramphenicol, utilized as a standard reference antibiotic, exhibited enhanced inhibition against all bacterial strains, hence validating the assay's trustworthiness. These results align with the established categorization standards for disc diffusion experiments, where inhibitory zones are categorized as follows: Inactive ( $\leq$  6 mm), weak activity (7 - 10 mm), moderate activity (11 - 15 mm), and strong activity ( $\geq$  16 mm) [28].

The MIC and MBC values (Table 3) confirmed these results. The methanol extract exhibited reduced MICs (31.25 - 250 ppm) against the majority of Gram-positive bacteria, especially *L. monocytogenes* (MIC: 31.25 ppm), indicating significant bacteriostatic or

bactericidal efficacy. The MBC/MIC ratio for *L. monocytogenes*, *B. subtilis*, and *S. aureus* was  $\leq$  4, signifying a bactericidal mechanism of action. The ratio for *E. coli* and *V. alginolyticus* surpassed 4, indicating a bacteriostatic effect. The antibacterial effects are likely due to polar metabolites in the methanol extract, including polyphenols, alkaloids, and flavonoids, which are known to damage bacterial membranes or block metabolic processes [29].

The antifungal activity of both extracts was tested against *Candida albicans*, *C. glabrata* and *C. krusei* using the Kirby-Bauer disc diffusion method and MIC/MFC assays. As shown in Table 2, methanol extract exhibited moderate activity against all three strains (zones of 12.35 - 15.54 mm), whereas the *n*-hexane extract showed no activity against *C. albicans* and *C. glabrata* ( $\leq$  6 mm), and weak activity (10.92 mm) against *C. krusei*.

**Table 2** Antifungal activities of various extracts from *B. rhodina* at concentrations 1,000 ppm.

No.	Microorganism	Clear zone diameter (mm)		
		<i>n</i> -Hexane extract	Methanol extract	Ketoconazole 30 ppm
1	<i>C. albicans</i>	6.71 $\pm$ 0.32 <sup>c</sup>	12.35 $\pm$ 0.65 <sup>b</sup>	24.86 $\pm$ 0.62 <sup>a</sup>
2	<i>C. glabrata</i>	6.62 $\pm$ 0.26 <sup>c</sup>	13.15 $\pm$ 0.32 <sup>b</sup>	23.42 $\pm$ 0.68 <sup>a</sup>
3	<i>C. krusei</i>	10.92 $\pm$ 0.18 <sup>c</sup>	15.54 $\pm$ 0.65 <sup>b</sup>	25.67 $\pm$ 0.63 <sup>a</sup>

The results were expressed as mean  $\pm$  Standard Deviation (SD) in the same row with different superscript letters indicates a statistically significant difference based on Tukey's test ( $\alpha = 5\%$ ).

**Table 3** The minimum inhibition concentration (MIC) and minimum bactericidal concentration (MBC)/ minimum fungicidal concentration (MFC) of various extracts from *B. rhodina*.

No.	Microorganism	MIC (ppm)			MBC (ppm)		
		<i>n</i> -Hexane extract	Methanol extract	Chloramphenicol	<i>n</i> -Hexane extract	Methanol extract	Chloramphenicol
1	<i>B. cereus</i>	500	250	31.25	1,000	500	62.5
2	<i>B. subtilis</i>	250	125	31.25	500	500	62.5
3	<i>E. coli</i>	>1,000	1000	31.25	>1,000	>1,000	62.5
4	<i>L. monocytogenes</i>	>1,000	31.25	31.25	>1,000	62.5	62.5
5	<i>S. typhimurium</i>	500	250	31.25	1,000	1,000	62.5
6	<i>S. aureus</i>	250	250	31.25	500	500	62.5
7	<i>V. alginolyticus</i>	>1,000	62.5	31.25	>1,000	125	125
8	<i>V. parahaemolyticus</i>	500	250	31.25	1,000	500	125

No.	Microorganism	MIC (ppm)			MFC (ppm)		
		<i>n</i> -Hexane extract	Methanol extract	Ketoconazole	<i>n</i> -Hexane extract	Methanol extract	Ketoconazole
1	<i>C. albicans</i>	1,000	500	15.62	>1,000	1,000	31.25
2	<i>C. glabrata</i>	1,000	500	15.62	>1,000	1,000	31.25
3	<i>C. krusei</i>	500	250	15.62	>1,000	500	31.25

The MIC values (**Table 3**) corroborated these results, indicating that the methanol extract exhibited MICs between 250 and 500 ppm, and MFCs from 500 - 1,000 ppm. Notably, all MFC/MIC ratios above 4, signifying a fungistatic mechanism of action. The methanol extract had significant effectiveness, particularly against *C. krusei*, which frequently shows resistance to standard azoles, although it is less potent than ketoconazole (MIC: 15.62 ppm). The findings indicate that *B. rhodina* synthesizes secondary compounds with prospective antifungal uses, especially against opportunistic pathogenic *Candida* species.

Although the antifungal activity of *B. rhodina* extracts was moderate compared to their antibacterial potency, this difference may be attributed to fundamental structural and physiological disparities between bacterial and fungal cells. The chitin-glucan matrix and ergosterol-enriched membranes of fungi typically limit the permeability of polar metabolites, such as phenolics and flavonoids, that predominated in the methanol extract. Consequently, metabolites that act effectively on bacterial cell walls or membrane systems may exhibit reduced interaction with fungal targets. The dominance of compounds like kojic acid, glycitein, and fraxetin—well-known for strong antibacterial but only moderate antifungal activities—suggests a metabolite composition biased toward antibacterial functionality.

Nevertheless, the measurable inhibition of *C. krusei*, a strain often resistant to conventional azole antifungals, indicates that certain metabolites in the extract may exert selective fungistatic activity. Such effects are likely associated with interference in ergosterol biosynthesis or disruption of fungal redox balance, leading to partial growth suppression rather than fungicidal action. The high MFC/MIC ratios ( $\geq 4$ ) further support this interpretation. Importantly, this outcome also implies that under modified culture conditions or co-culture stimulation, *B. rhodina* could potentially activate latent biosynthetic gene clusters

responsible for more potent antifungal metabolites. Future bioassay-guided fractionation and transcriptomic analysis are warranted to elucidate the structures and biosynthetic pathways of these compounds.

Overall, while the antifungal effect is moderate, the observed selectivity underscores *B. rhodina*'s adaptive metabolic potential. The findings suggest a biosynthetic preference toward antibacterial secondary metabolism while maintaining a baseline antifungal capability that could be enhanced through fermentation optimization or elicitation strategies. This dual-spectrum potential positions *B. rhodina* as a valuable candidate for future antimicrobial drug discovery efforts.

The antibacterial efficacy demonstrated in this study for *B. rhodina* corresponds with and considerably enhances previous findings about the bioactivity of *Botryosphaeria* species. In the restricted yet significant literature, Abdou *et al.* [30] presented the inaugural complete evidence of antimicrobial metabolite synthesis from *B. rhodina*, isolating and assessing two depsidone compounds—Botryorhodine A and Botryorhodine B—for their antifungal and cytotoxic properties. [30]. Botryorhodine A displayed a MIC of 26.03  $\mu\text{M}$  against *A. terreus* and 191.60  $\mu\text{M}$  against *Fusarium oxysporum*, whereas Botryorhodine B showed MICs of 49.70 and 238.80  $\mu\text{M}$  against the same fungi, respectively. [30] This study extends previous findings by showing that crude extracts, especially the methanol-soluble fraction, of *B. rhodina* cultivated under solid-state fermentation conditions display significant and extensive antibacterial properties. The methanol extract had significant inhibitory effects against *L. monocytogenes* (28.20 mm), *B. subtilis* (21.37 mm), and *S. aureus* (18.41 mm), along with modest antifungal activity against *Candida* species. Although our study did not isolate or quantify specific compounds like Botryorhodines, the observed antimicrobial activities—especially those demonstrating bactericidal or

fungistatic properties indicated by MIC/MBC and MIC/MFC ratios—imply the existence of structurally analogous or similarly functioning metabolites.

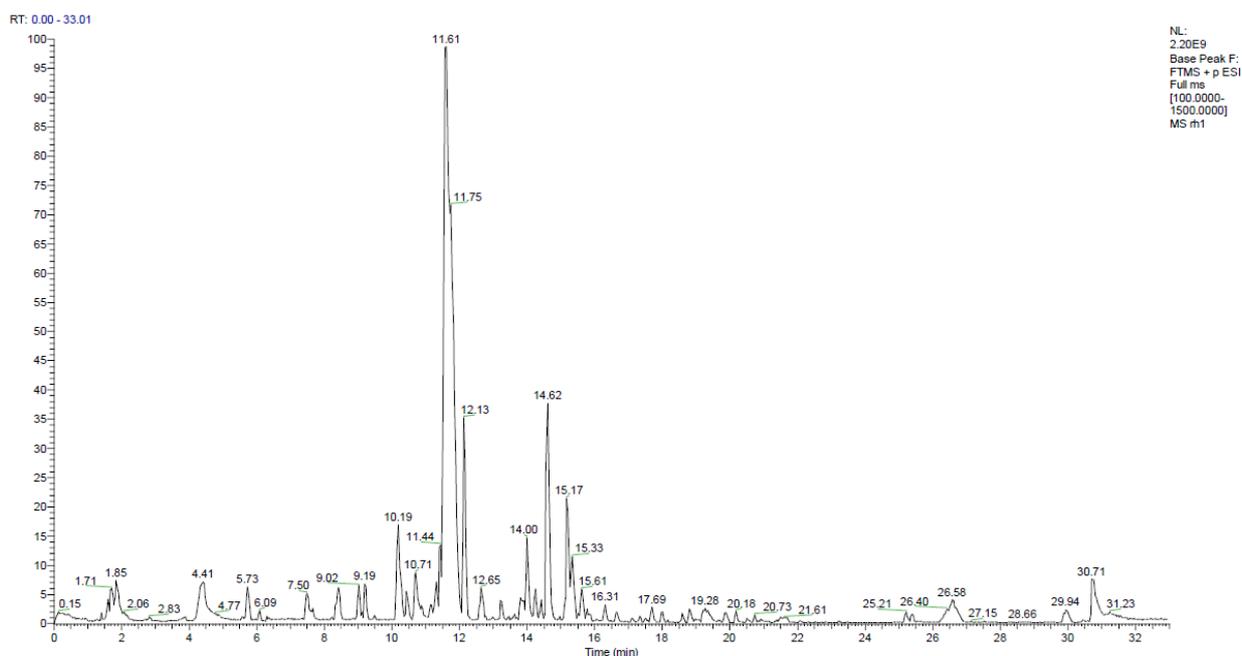
Aside from *B. rhodina*, documentation regarding other Botryosphaeria species exhibiting antibacterial characteristics is scarce. Karličić *et al.* [31] evaluated the inhibition of mycelial growth of *B. dothidea*, *Dothiorella sarmentorum*, and *Neofusicoccum parvum* under the biocontrol influence of *Trichoderma* species and pine bark extracts. Nonetheless, this research concentrated mostly on antagonistic interactions instead of metabolite-mediated antibacterial efficacy. For example, dual-culture tests with *Trichoderma* achieved 67% - 85% mycelial suppression of *B. dothidea*, but pine bark extracts elicited only 39% - 44% inhibition. No antimicrobial chemicals were directly extracted from *B. dothidea*, highlighting the distinctiveness of *B. rhodina* as a biosynthetic reserve within the genus. [31].

The ecological significance of *B. rhodina* as an endophyte is likewise remarkable. Endophytic fungi are recognized for their role in chemical defense for host plants, synthesizing antimicrobial secondary metabolites that inhibit the proliferation of competing

microbial diseases within the plant. This mutualistic characteristic has been suggested as a catalyst in the evolution of varied and bioactive metabolite biosynthetic gene clusters in endophytes. The presence of *B. rhodina* in mangrove environments, marked by variable salinity, tidal stress, and intense microbial competition, may enhance the synthesis of structurally distinct and bioactive chemicals.

#### Metabolites profiling of *B. rhodina* extracts

Metabolite profiling revealed a chemically diverse set of secondary metabolites, highlighting the fungus's metabolic potential. A broad array of both polar and non-polar metabolites, encompassing various chemical classes, was identified using complementary analytical platforms: LC-MS/MS for the methanol extract and GC-MS for the n-hexane extract. **Figure 4** depicts the Total Ion Chromatogram (TIC) of the methanol extract derived from UHPLC-Q-Orbitrap LC-MS/MS analysis, showcasing an intricate metabolite profile characterized by several high-intensity peaks over the retention time spectrum. This indicates the significant chemical variety and polarity of the extract.



**Figure 4** Total Ion Chromatogram (TIC) from a High-Resolution Mass Spectrometry (HRMS) of methanol extract.

**Figure 4** presents the Total Ion Chromatogram (TIC) derived from high-resolution LC-MS/MS analysis of the methanol extract of *B. rhodina*. The TIC displays

an intricate and concentrated distribution of peaks within the 0-32-minute retention time interval, indicating the existence of a chemically varied

assortment of polar and semi-polar secondary metabolites. The elevated signals detected between 5 and 25 min indicate the elution of numerous mid-polarity chemicals, such as flavonoids, phenolics, alkaloids, and polyketides, many of which have been previously linked to antibacterial properties [27]. The numerous peaks signify a substantial biosynthetic yield

under the specified solid-state fermentation conditions, underscoring the metabolic abundance of the extract. This chromatographic profile underpins the subsequent identification of compounds detailed in **Table 4** and corroborates the extensive antibacterial capability demonstrated in the biological experiments.

**Table 4** Metabolite profiling of methanol extract.

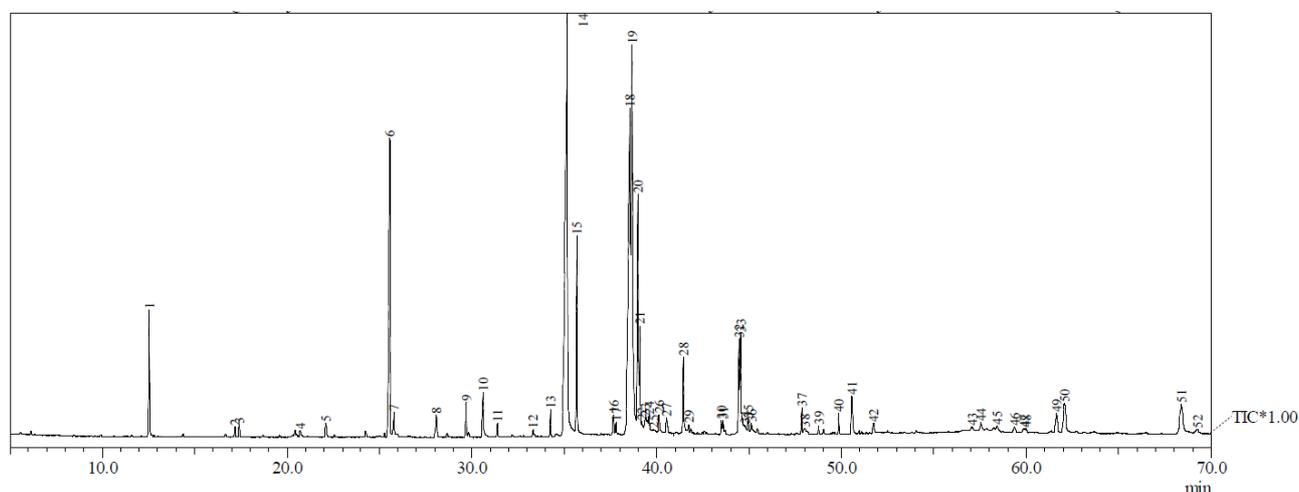
No	Name	Formula	Annot. DeltaMass [ppm]	Calc. MW	Area
1	Kojic acid	C <sub>6</sub> H <sub>6</sub> O <sub>4</sub>	- 1.9	142.02634	611238473.7
2	4-Acetamidobutanoic acid	C <sub>8</sub> H <sub>11</sub> NO <sub>3</sub>	- 1.32	145.0737	1449448289
3	Pyridoxal	C <sub>8</sub> H <sub>9</sub> NO <sub>3</sub>	- 15.52	167.05565	338653086.6
4	(Hydroxyethyl)methacrylate	C <sub>6</sub> H <sub>10</sub> O <sub>3</sub>	- 0.85	130.06288	1736773156
5	Sorbic acid	C <sub>6</sub> H <sub>8</sub> O <sub>2</sub>	0.69	112.05251	357401798.7
6	Unknown 23	C <sub>8</sub> H <sub>9</sub> NO <sub>2</sub>	- 1.33	151.06313	274206821.4
7	N-Acetylvaline	C <sub>7</sub> H <sub>13</sub> NO <sub>3</sub>	- 1.36	159.08933	393009948.7
8	OI1700000	C <sub>7</sub> H <sub>12</sub> O <sub>3</sub>	- 1.26	144.07846	700728707.4
9	LV1850000	C <sub>7</sub> H <sub>8</sub> O <sub>3</sub>	- 2.14	140.04704	1058266413
10	N-(2-OXOOXOLAN-3-YL)HEXANAMIDE	C <sub>10</sub> H <sub>17</sub> NO <sub>3</sub>	- 1.49	199.12055	329035257
11	Phenylethyl alcohol	C <sub>8</sub> H <sub>10</sub> O	0.59	122.07324	445092318.9
12	Vanillyl alcohol	C <sub>8</sub> H <sub>10</sub> O <sub>3</sub>	- 1.12	154.06282	412503890.6
13	2-Acetylcyclohexanone	C <sub>8</sub> H <sub>12</sub> O <sub>2</sub>	- 1.87	140.08347	755270815.2
14	Ethyl 3-oxohexanoate	C <sub>8</sub> H <sub>14</sub> O <sub>3</sub>	- 0.82	158.09416	395148814.1
15	DL-2-(acetylamino)-3-phenylpropanoic acid	C <sub>11</sub> H <sub>13</sub> NO <sub>3</sub>	- 0.6	207.08942	471678300.2
16	N-(2-furylmethylidene)-(4-((2-furylmethylidene)amino)methyl)cyclohexyl)methanamine	C <sub>18</sub> H <sub>22</sub> N <sub>2</sub> O <sub>2</sub>	- 1.5	298.16768	1250462144
17	Dodecanedioic acid	C <sub>12</sub> H <sub>22</sub> O <sub>4</sub>	- 1.65	230.15143	820355426.9
18	NP-001501	C <sub>16</sub> H <sub>17</sub> N <sub>3</sub> O <sub>2</sub>	- 1.98	283.13152	255969460
19	NP-021844	C <sub>12</sub> H <sub>24</sub> O <sub>5</sub>	88576.73	270.14379	274570720.2
20	(2S,3S)-2-hydroxytridecane-1,2,3-tricarboxylic acid	C <sub>16</sub> H <sub>28</sub> O <sub>7</sub>	- 2.06	332.18282	305562642.3
21	4,4'-Bis(diethylamino)benzophenone	C <sub>21</sub> H <sub>28</sub> N <sub>2</sub> O	- 2.61	324.21932	2888445537
22	Tetraneurin A	C <sub>17</sub> H <sub>22</sub> O <sub>6</sub>	- 1.86	322.14104	259063397.3
23	Unknown 1	C <sub>23</sub> H <sub>30</sub> N <sub>2</sub> O <sub>3</sub>	- 1.54	382.22505	710625326.5
24	Unknown 2	C <sub>22</sub> H <sub>20</sub> N <sub>4</sub> O <sub>5</sub>	- 2.42	420.14235	274065787.1
25	Ofloxacin methyl ester	C <sub>19</sub> H <sub>22</sub> FN <sub>3</sub> O <sub>4</sub>	- 5.43	375.1574	725248196.1
26	MFCD23380831	C <sub>16</sub> H <sub>13</sub> NO <sub>6</sub>	- 1.81	315.07372	696943477.5
27	Unknown 3	C <sub>16</sub> H <sub>15</sub> NO <sub>7</sub>	- 2.46	333.08403	365139587.8
28	Unknown 4	C <sub>17</sub> H <sub>17</sub> NO <sub>7</sub>	- 2.43	347.09966	416699417.5

No	Name	Formula	Annot. DeltaMass [ppm]	Calc. MW	Area
29	Erlotinib	C <sub>22</sub> H <sub>23</sub> N <sub>3</sub> O <sub>4</sub>	- 1.75	393.16817	1011283886
30	Unknown 5	C <sub>22</sub> H <sub>28</sub> O <sub>2</sub> S	- 2.53	356.1801	332104504.3
31	Vesnarinone	C <sub>22</sub> H <sub>25</sub> N <sub>3</sub> O <sub>4</sub>	- 1.26	395.18401	265101829.3
32	NP-021781	C <sub>19</sub> H <sub>36</sub> O <sub>5</sub>	63828.58	366.22966	52316910304
33	Unknown 6	C <sub>40</sub> H <sub>72</sub> N <sub>4</sub> S <sub>5</sub>	2.11	768.43767	298424562.1
34	(3,4-Dimethoxyphenyl)((3S)-3-[5-(3-pyridinyl)-1,3,4-oxadiazol-2-yl]-1-pyrrolidinyl} methanone	C <sub>20</sub> H <sub>20</sub> N <sub>4</sub> O <sub>4</sub>	57827.89	402.13164	5394895666
35	2-(3,4-Dimethoxyphenyl)-5,7,8-trihydroxy-3-methoxy-2,3-dihydro-4H-chromen-4-one	C <sub>18</sub> H <sub>18</sub> O <sub>8</sub>	- 2.59	362.09923	260749886.7
36	Unknown 7	C <sub>21</sub> H <sub>28</sub> N <sub>2</sub>	- 2.42	308.2245	648233913.7
37	Geranyl glucoside	C <sub>16</sub> H <sub>28</sub> O <sub>6</sub>	- 2.23	316.18788	887106925.6
38	Octyl (4xi)-2-deoxy-2-[(Z)-(1-hydroxyethylidene)amino]-alpha-D-xylo-hexopyranoside	C <sub>16</sub> H <sub>31</sub> NO <sub>6</sub>	- 2.48	333.21431	386956579.3
39	Unknown 8	C <sub>14</sub> H <sub>22</sub> N <sub>6</sub> O <sub>4</sub>	- 1.65	338.16969	351044651.8
40	(1S,3S,12S,14R,27S)-12-Hydroxy-1,3-dimethyl-2,5,15,23,25-pentaazaheptacyclo[12.10.2.1~2,5~.0~6,11~.0~12,27~.0~15,24~.0~17,22~]heptacosane-6,8,10,17,19,21,23-heptaene-4,16,26-trione	C <sub>24</sub> H <sub>21</sub> N <sub>5</sub> O <sub>4</sub>	- 1.66	443.15862	631153618.7
41	4-Methoxycinnamic acid	C <sub>10</sub> H <sub>10</sub> O <sub>3</sub>	- 1.65	178.0627	961957815.5
42	Deoxybrevianamide E	C <sub>21</sub> H <sub>25</sub> N <sub>3</sub> O <sub>2</sub>	- 1.97	351.19399	263163030.8
43	9-Methoxy-12-(2-methyl-1-propen-1-yl)-1,2,3,5a,6,11,12,14a-octahydro-5H,14H-pyrrolo[1'',2'':4',5']pyrazino[1',2':1,6]pyrido[3,4-b]indole-5,14-dione	C <sub>22</sub> H <sub>25</sub> N <sub>3</sub> O <sub>3</sub>	- 2.41	379.18868	1657865510
44	Unknown 9	C <sub>18</sub> H <sub>32</sub> O <sub>6</sub>	- 1.84	344.21926	419291370.8
45	PPG Acrylate n5	C <sub>18</sub> H <sub>34</sub> O <sub>7</sub>	60682.5	384.2115	254965891.2
46	Usnic acid	C <sub>18</sub> H <sub>16</sub> O <sub>7</sub>	- 2.44	344.08876	509685447.5
47	Unknown 10	C <sub>38</sub> H <sub>24</sub> N <sub>8</sub> O <sub>6</sub>	- 3.74	688.17931	500753787.2
48	NP-005405	C <sub>17</sub> H <sub>16</sub> O <sub>5</sub>	- 1.41	300.09935	300686318.2
49	Unknown 11	C <sub>24</sub> H <sub>40</sub> O <sub>8</sub>	- 1.66	456.27156	412576947.9
50	3,4-Dihydroxyphenylpropionic acid	C <sub>9</sub> H <sub>10</sub> O <sub>4</sub>	- 98929.48	164.04701	260039679.7
51	Fraxetin	C <sub>10</sub> H <sub>8</sub> O <sub>5</sub>	- 2.72	208.03661	3464812649

No	Name	Formula	Annot. DeltaMass [ppm]	Calc. MW	Area
52	5,7-DIHYDROXY-3',4',5'-TRIMETHOXYFLAVANONE	C <sub>18</sub> H <sub>18</sub> O <sub>7</sub>	- 2.41	346.10442	3193661421
53	N-(4-fluorophenyl)-2-methyl-5-(morpholin sulfonyl)-3-furamide	C <sub>16</sub> H <sub>17</sub> FN <sub>2</sub> O <sub>5</sub> S	5.95	368.08641	341196330.6
54	Glycitein	C <sub>16</sub> H <sub>12</sub> O <sub>5</sub>	- 2.27	284.06783	2008325644
55	[(5aS,6S,7aR,8R,9S,11aS,11bS,12R)-6,9-bis(acetyloxy)-12-hydroxy-5a,8,11a-trimethyl-1-oxo-3-(pyridin-3-yl)-1,5a,6,7,7a,8,9,10,11,11a,11b,12-dodecahydro-2,5-dioxatetraphen-8-yl]methyl acetate	C <sub>31</sub> H <sub>37</sub> NO <sub>10</sub>	- 1.25	583.24101	345610334.8
56	Unknown 12	C <sub>21</sub> H <sub>19</sub> NO <sub>7</sub>	- 2.04	397.11534	261918464.7
57	Unknown 13	C <sub>22</sub> H <sub>36</sub> N <sub>6</sub> O <sub>7</sub>	- 0.75	496.26418	327918171.7
58	Unknown 14	C <sub>24</sub> H <sub>42</sub> O <sub>9</sub>	- 1.6	474.28212	919030640.6
59	Unknown 15	C <sub>22</sub> H <sub>21</sub> NO <sub>7</sub>	- 2.62	411.13072	399419961.5
60	Unknown 16	C <sub>24</sub> H <sub>20</sub> N <sub>4</sub> O <sub>5</sub>	- 2.09	444.14244	339186996.8
61	Unknown 17	C <sub>32</sub> H <sub>56</sub> O <sub>12</sub>	- 3.46	632.37499	540939073.8
62	Unknown 18	C <sub>36</sub> H <sub>56</sub> O <sub>4</sub> P <sub>2</sub>	- 0.96	614.36479	525531675
63	Unknown 19	C <sub>40</sub> H <sub>70</sub> O <sub>15</sub>	- 3.46	790.46874	410030023.4
64	Unknown 20	C <sub>27</sub> H <sub>31</sub> N <sub>3</sub> O <sub>4</sub>	- 2.37	461.23036	453586050.2
65	Unknown 21	C <sub>30</sub> H <sub>55</sub> N <sub>5</sub> O <sub>5</sub>	- 3.1	565.41857	457024359.2
66	Unknown 22	C <sub>42</sub> H <sub>74</sub> O <sub>15</sub>	- 2.43	818.50078	359907611.2
67	Oleamide	C <sub>18</sub> H <sub>35</sub> NO	- 1.4	281.27147	323460282.1

Similarly, **Figure 5** illustrates the TIC obtained from GC-MS analysis of the *n*-hexane extract, with a more confined distribution of peaks primarily within the mid to late retention time range, signifying the presence

of less polar chemicals such as fatty acids, esters, and sterols. The chromatograms show distinct chemical compositions for each extract, reflecting differences in solvent polarity and extraction efficiency.



**Figure 5** Total Ion Chromatogram (TIC) from GC-MS analysis of the *n*-hexane extract.

**Figure 5** presents the Total Ion Chromatogram (TIC) derived from the GC-MS analysis of the *n*-hexane extract of *B. rhodina*, illustrating a comparatively simpler chemical profile than that of the methanol extract. The majority of peaks are concentrated in the 20-45-minute retention time range, corresponding primarily to the elution of non-polar lipophilic compounds such as saturated and unsaturated fatty acids, esters, and sterols. The intensity and distribution of these peaks suggest the extract is dominated by

compounds with limited polarity, such as ethyl palmitate, linoleic acid, and clionasterol, which are known to exhibit mild to moderate antimicrobial or anti-inflammatory properties. Although less complex than the methanolic TIC, this chromatographic pattern provides valuable insight into the lipid-based metabolome of *B. rhodina*, as detailed in **Table 5**, and supports the differential antimicrobial activities observed between polar and non-polar extracts.

**Table 5** Metabolite profiling of *n*-Hexane extract.

No	Name	Area
1	Phenylethyl Alcohol	16342713
2	Acetic acid, 2-phenylethyl ester	1253967
3	5-Hydroxy-2-Decenoic Acid Lactone	1689771
4	Benzaldehyde, 4-ethoxy-	1020155
5	Benzeneethanol, 4-hydroxy-	2064942
6	Dodecanoic acid	4913178
7	Benzaldehyde, 4-ethoxy- (CAS) p-Ethoxybenzaldehyde	4355374
8	Benzaldehyde, 3-ethoxy-	5424632
9	Tetradecanoic acid (CAS) Myristic acid	8398044
10	Tetradecanoic acid, ethyl ester (CAS) Ethyl myristate	1677370
11	1H-2-Benzopyran-1-one, 3,4-dihydro-3,8-dihydroxy-3-methyl-, (-)-	1572026
12	Hexadecanoic acid, methyl ester	3373432
13	Hexadecanoic acid, ethyl ester (CAS) Ethyl palmitate	25153720
14	9,12-Hexadecadienoic acid, methyl ester (CAS) METHYL-9,12-Hexadecadienoate	3361719
15	11-Octadecenoic acid, methyl ester, (Z)-	1833382
16	Ethyl Oleate	13861020
17	9-Octadecenoic acid (Z)-, ethyl ester (CAS) Ethyl oleate	2515482
18	9,12-Octadecadienoic acid (Z,Z)- (CAS) Linoleic acid	6227145
19	Octadecanoic acid, ethyl ester	3220091
20	9,12-Octadecadienoic acid (Z,Z)- (CAS) Linoleic acid	1215008
21	9,12-Octadecadienoic acid (Z,Z)-	3597971
22	Isoamyl laurate	2702163
23	Octadecanoic acid, 2-hydroxy-1,3-propanediyl ester	11995847
24	8,11-Octadecadienoic acid, methyl ester	1465620
25	9,11-Octadecadienoic acid, methyl ester, (E,E)-	2112308
26	9-Octadecene, 1-methoxy-, (E)- (CAS) 9(TRANS)-OCTADEC-1-Enyl Methyl Ether	2199505
27	9,12-Octadecadien-1-ol (CAS) OCTADECA-9,12-DIEN-1-OL	14524843
28	DI-(9-OCTADECENOYL)-GLYCEROL	20306995
29	Octadecanoic acid, 3-oxo-, methyl ester (CAS) Methyl 3-oxooctadecanoate	1831249
30	Octadecanoic acid, 2-hydroxy-1,3-propanediyl ester	2872009
31	Hexadecanoic acid, 2-hydroxy-1-(hydroxymethyl)ethyl ester	3338067
32	Octanoic acid, 2-phenylethyl ester	3736257

No	Name	Area
33	9-Octadecenoic acid (Z)-, 2,3-dihydroxypropyl ester	1965918
34	Di-n-octyl phthalate	1192388
35	2,6,10,14,18,22-Tetracosahexaene, 2,6,10,15,19,23-hexamethyl- (CAS) Squalene	2605876
36	Benzene, 1-(1-cyclohexen-1-yl)ethyl -	8112199
37	9-Octadecyne	2297577
38	Octadecanoic acid, ethenyl ester (CAS) Vinyl stearate	2103125
39	Stigmast-5-en-3-ol, (3.beta.,24S)- (CAS) Clionasterol	1604069
40	Stigmast-4-en-3-one	1537555
41	Ergosta-4,7,22-trien-3-one	1143670
42	4,22-Cholestadien-3-one	1001862
43	Stigmast-4-en-3-one	5911489
44	(R)-(-)-14-Methyl-8-hexadecyn-1-ol	12119013

Compound identification was based on tandem mass spectrometric (MS<sup>2</sup>) data acquired using high-resolution UHPLC-Q Exactive Orbitrap LC-MS/MS and validated through comparison with mzCloud, ChemSpider, and MassBank databases at a mass tolerance of  $\pm 5$  ppm. Fragmentation spectra were matched to reference libraries to enhance annotation reliability. Therefore, metabolites such as kojic acid, usnic acid, and glycitein were putatively identified based on accurate mass, retention time, and high spectral similarity, pending further structural confirmation by NMR.

The PLS-DA models generated from LC-MS/MS (**Figure 5**) and GC-MS (**Figure 6**) datasets demonstrated clear separation between methanolic and hexane extracts, confirming distinct metabolomic profiles. VIP score plots indicated that polar metabolites, including kojic acid, fraxetin, and glycitein, were major contributors to the antibacterial activity, while several uncharacterized metabolites with high VIP values suggest the presence of novel bioactive chemical entities requiring future isolation and NMR confirmation.

The utilization of high-resolution Orbitrap-based liquid chromatography-mass spectrometry (LC-MS/MS) enabled the thorough detection and provisional identification of 67 unique secondary metabolites in the methanol extract of *B. rhodina*. This analytical method provided outstanding mass accuracy, elevated resolving power, and sensitive detection of a wide range of polar and semi-polar molecules, facilitating precise molecular feature extraction and annotation [32]. The utilization of

tandem mass spectrometry and spectrum database matching through ChemSpider and mzCloud platforms greatly boosted chemical identification and structural prediction. Simultaneously, gas chromatography-mass spectrometry (GC-MS) was utilized to examine the *n*-hexane extract, which is rich in non-polar components. This method facilitated the identification of 45 volatile and semi-volatile constituents, including fatty acids, esters, and sterols, according to their retention times and fragmentation patterns. The combined orthogonal metabolomic platforms yielded a thorough chemical profile of both polar and non-polar fractions, as detailed in **Tables 4** and **5**, therefore clarifying the metabolic complexity and biochemical potential of this mangrove-derived endophytic fungus.

The methanol extract contained diverse compounds, including phenolics, flavonoids, alkaloids, terpenoids, and heterocyclic derivatives. Significant levels of well-documented antimicrobial metabolites, including kojic acid, usnic acid, glycitein, fraxetin, vanillyl alcohol, and deoxybrevianamide E were —were putatively identified based on accurate mass and MS/MS spectral similarity to reference spectra in the mzCloud and ChemSpider databases. These annotations are considered tentative and require further structural confirmation through isolation and NMR analysis. [33-37]. These chemicals are extensively acknowledged in the literature for their antibacterial, antifungal, or cytoprotective properties, indicating a functional role in the bioactivity identified in the extract. The existence of 25 unidentified compounds with significant relative abundance, including those with intricate nitrogen-

containing or halogenated structures, underscores the potential for novel bioactive substances. Future investigations should focus on the structural elucidation of these unidentified compounds using bioassay-guided fractionation and NMR spectroscopy.

The discussion generally focuses on antibacterial action; however, numerous metabolites found in the methanol extract, including kojic acid, usnic acid, and vanillyl alcohol, are recognized for their antifungal characteristics as well [33,34,36]. These chemicals may facilitate the mild inhibition noted against *Candida* species, especially *C. krusei*, which frequently demonstrates resistance to standard antifungals. Additionally, certain undiscovered metabolites with elevated peak intensities may possess new antifungal scaffolds. The lack of isolated targeted antifungal compounds restricts accurate attribution of bioactivity; however, the findings suggest that the methanol extract encompasses multifunctional secondary metabolites with potential antibacterial and antifungal applications. Future research utilizing bioassay-guided purification and fungal-specific synergy testing is crucial to clarify these connections.

In contrast, the *n*-hexane extract primarily comprised saturated and unsaturated fatty acids (e.g., dodecanoic acid, myristic acid, linoleic acid), fatty acid esters (e.g., ethyl palmitate, ethyl oleate), sterols (e.g., clionasterol), and long-chain hydrocarbons. While these constituents are typically regarded as less effective direct antimicrobial agents than their polar counterparts, some have been documented to possess mild antibacterial activities and may synergistically enhance the overall activity profile. [38].

The results combined indicate a substantial segregation of metabolites that is dependent on polarity. The methanol extract, comprising a chemically varied and pharmacologically important collection of polar molecules, exhibited superior and more extensive antibacterial activity compared to the *n*-hexane extract, which was predominantly limited to lipid-based constituents. This distinction provides a molecular rationale for the observed variations in bioactivity and underscores the importance of solvent selection in optimizing the extraction of secondary metabolites. The observations support the view that *B. rhodina* is a promising source of chemically diverse secondary metabolites, with significant potential for discovering

novel antimicrobial drugs from underexplored fungal endophytes in mangrove ecosystems.

An in-depth analysis of the methanol extract identified multiple well-defined bioactive chemicals exhibiting various antibacterial actions. Kojic acid, a secondary metabolite synthesized by several fungi, is recognized for its antibacterial and antifungal properties via metal ion chelation and inhibition of tyrosinase [39]. Usnic acid, a dibenzofuran derivative, demonstrates antibacterial properties through membrane disruption and the suppression of oxidative phosphorylation [34]. Flavonoid derivatives, including glycitein, fraxetin, and 5,7-dihydroxy-3',4',5'-trimethoxyflavone, exhibit antibacterial, antioxidant, and anti-inflammatory activities, typically through mechanisms such as efflux pump blockage, oxidative stress induction, and cell wall disruption [40]. Their simultaneous presence in the extract indicates possible synergistic interactions. Other notable chemicals comprise phenylethyl alcohol, a quorum sensing antagonist; vanillyl alcohol, which disrupts ergosterol biosynthesis; and pyridoxal, a vitamin B6 derivative that amplifies oxidative stress in microbial cells. The inclusion of nitrogen-containing heterocycles and polyketide-like structures, such as deoxybrevianamide E and tetranurin A, enhances the bioactive potential of the extract.

The antibacterial activity of the methanol extract was substantially linked with the concentration of these polar metabolites. Significant inhibition zones and low MIC values—31.25 ppm for *L. monocytogenes* and 125 ppm for *B. subtilis*—demonstrate bactericidal property. The detected antibacterial activity probably arises from a combination of individual and synergistic effects among these substances. Conversely, the *n*-hexane extract, characterized by fatty acids, esters, and sterols, demonstrated minimal antibacterial efficacy. Although certain elements, such as linoleic acid and methyl palmitate, have moderate membrane-disruptive effects, their total impact is quite insignificant.

Alongside the well-characterized chemicals, the identification of more than 25 unidentified metabolites in the methanol extract—many exhibiting intricate, high-molecular-weight structures—indicates the potential presence of novel chemical scaffolds with bioactive properties. These metabolites exhibited elevated peak intensities yet lacked conclusive matches in spectrum libraries, highlighting the constraints of

existing databases and the distinctive chemical space possessed by *B. rhodina*.

The correlation between metabolite chemistry and biological activity suggests that several compounds act through distinct and complementary mechanisms. Kojic acid likely exerts antibacterial effects via chelation of essential metal ions and inhibition of tyrosinase-dependent redox enzymes, leading to oxidative imbalance. Usnic acid, a dibenzofuran derivative, disrupts microbial membrane integrity and inhibits oxidative phosphorylation, accounting for its broad-spectrum activity. Flavonoid derivatives such as glycitein and fraxetin may inhibit efflux pumps, interfere with nucleic acid synthesis, and induce reactive oxygen species accumulation, thereby enhancing bactericidal efficacy. The presence of deoxybrevianamide E and nitrogen-containing polyketides further supports a multi-target mode of action involving redox modulation and membrane destabilization.

These observations indicate that the antimicrobial potential of *B. rhodina* extracts arises from a synergistic interplay of chemically diverse metabolites rather than a single dominant compound. Although the biosynthetic pathways responsible for these metabolites were not experimentally validated in this study, their activation during the stationary growth phase likely reflects secondary metabolism associated with nutrient limitation and stress response, as widely reported in filamentous fungi.

To avoid overstatement, the discussion of sustainability has been moderated to note that solid-state fermentation represents a promising approach for eco-efficient metabolite production, aligning with principles of green chemistry, but further life-cycle analysis would be required to substantiate environmental benefits.

These findings align with and expand upon prior reports concerning *B. rhodina* and related Botryosphaeriaceae. Abdou *et al.* [30] identified Botryorhodines A-D, depsidone-type compounds exhibiting antifungal properties, from cultures of *B. rhodina* [30]. Despite the absence of these particular chemicals in the current investigation, the detection of structurally analogous phenolics and polyketides, including fraxetin and deoxybrevianamide E, implies similar metabolic potential. Moreover, in contrast to previous research centered on targeted isolation, this

untargeted metabolomics methodology revealed a wider and more varied chemical profile, especially under solid-state fermentation conditions that might replicate the fungus's natural habitat.

The genus *Botryosphaeria* has been inadequately represented in natural product research, with the majority of publications concentrating on its function as a plant pathogen or in co-culture inhibition experiments [41]. The present investigation identifies *B. rhodina*, particularly from a mangrove environment, as a significant source of chemically unique and pharmacologically advantageous metabolites. Environmental stresses typical of the mangrove ecosystem may stimulate the synthesis of adaptive secondary metabolites rarely observed in terrestrial strains, underscoring the ecological importance of endophyte-derived bioactives [42].

In conclusion, the compositional complexity, presence of structurally varied and pharmacologically active metabolites, and potential synergistic interactions elucidate the enhanced antibacterial efficacy of the methanol extract. These findings highlight the importance of metabolite profiling in understanding the role of endophytic fungi and discovering antimicrobial agents from natural sources.

Notwithstanding the thorough metabolite profiling and antimicrobial evaluations conducted in this work, certain limitations must be recognized. The identification of metabolites using LC-MS/MS and GC-MS relied mostly on spectrum library matching and *in silico* fragmentation predictions, lacking confirmation structural elucidation using NMR or MS/MS fragmentation for the unidentified molecules. This creates a limitation on the reliability of chemical identification, especially for metabolites with unclear or non-existent database correlations.

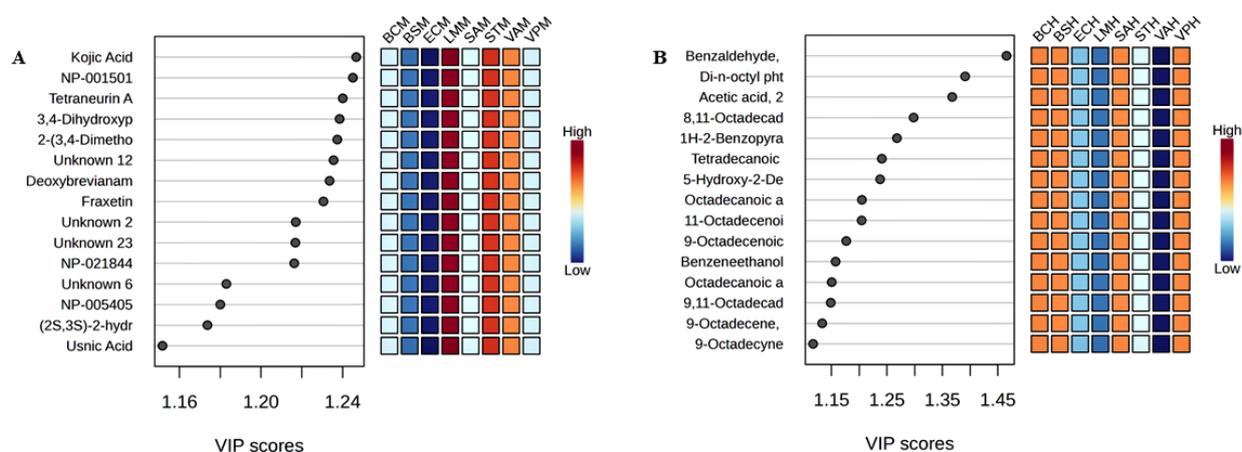
While this study successfully revealed a wide range of secondary metabolites from *Botryosphaeria rhodina* using untargeted LC-HRMS and GC-MS analyses, it is important to note that the compound identifications are putative and based solely on accurate mass and spectral database matching. Consequently, experimental evaluation of cytotoxicity or *in silico* pharmacokinetic assessment using tools such as SwissADME was not performed. Such predictive analyses require structurally confirmed compounds to ensure valid and reproducible results, whereas applying

them to tentative chemical identities could lead to speculative or misleading interpretations. Therefore, the current findings should be viewed as a foundational step toward understanding the metabolic potential of *B. rhodina*. Future investigations will focus on bioassay-guided isolation and structure elucidation of the key bioactive metabolites, followed by comprehensive cytotoxicity, ADME, and drug-likeness evaluations through both experimental and computational approaches to substantiate their therapeutic relevance and safety profile.

It should be noted that all metabolite annotations presented herein are considered putative identifications (MSI level 2), derived from high-resolution MS data and spectral database matching. Definitive structural confirmation was not performed and will be addressed in future work.

### Multivariate analysis (PLS-DA) of metabolite profiles

A Partial Least Squares Discriminant Analysis (PLS-DA) was conducted to quantify the connection between metabolite composition and antibacterial activity in the methanolic, and *n*-hexane extracts of *Botryosphaeria rhodina*. This chemometric method established a statistical framework to differentiate extract profiles based on metabolite composition and to find key discriminant metabolites that significantly contribute to the observed biological activities. The PLS-DA score plots (**Figure 6**) demonstrated a distinct separation between the methanolic (M) and *n*-hexane (H) extracts for all evaluated bacterial strains. The methanol extract exhibited a close clustering with *L. monocytogenes*, *S. aureus*, and *B. subtilis*, correlating with the most substantial inhibitory zones and the minimal MIC/MBC values seen in antimicrobial testing.



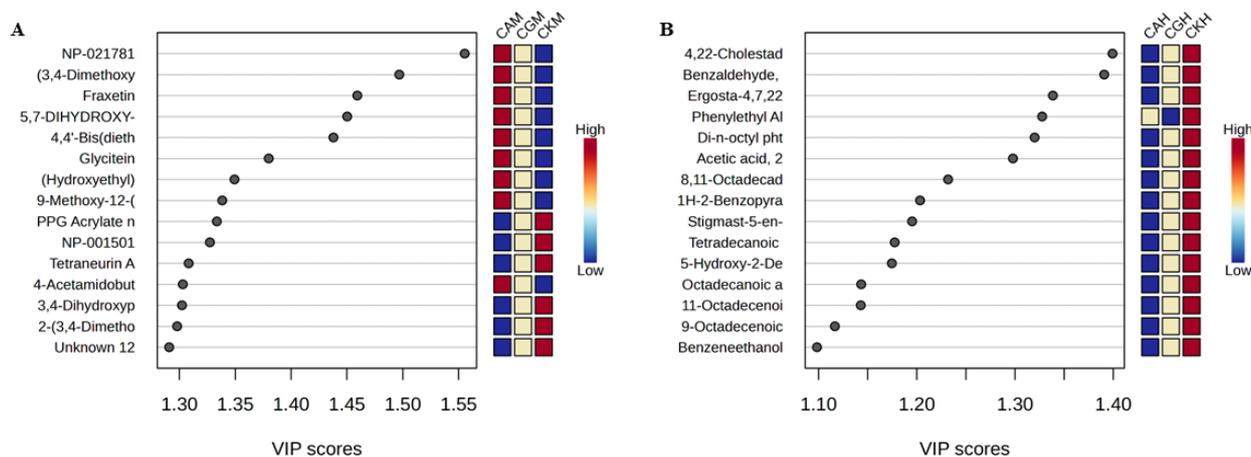
**Figure 6** PLS-DA of *Botryosphaeria rhodina* extracts against pathogenic bacteria. (A) Methanol extract (M); (B) *n*-Hexane extract (H). Heatmaps illustrate the relative abundance of discriminant metabolites across bacterial strains: BC (*Bacillus cereus*), BS (*Bacillus subtilis*), EC (*Escherichia coli*), LM (*Listeria monocytogenes*), ST (*Salmonella typhimurium*), SA (*Staphylococcus aureus*), VA (*Vibrio alginolyticus*), and VP (*Vibrio parahaemolyticus*). VIP scores highlight metabolites contributing most to class separation.

The metabolites exhibiting the greatest Variable Importance in Projection (VIP) scores comprised kojic acid, glycitein, fraxetin, and deoxybrevianamide E. These polar chemicals are extensively documented for their antibacterial actions, which encompass membrane disruption, enzyme inhibition, and the generation of oxidative stress. Their robust correlation with Gram-positive bacteria indicates that they are the primary agents of the observed bactericidal activity.

Conversely, the *n*-hexane extract exhibited diminished clumping and a looser association with *Escherichia coli* and *Vibrio* spp., aligning with its restricted antibacterial efficacy. The distinguishing metabolites in this fraction were primarily fatty acids, sterols, and long-chain hydrocarbons, which are often noted for their poor or inconsistent antibacterial activity and contributed little to class differentiation.

PLS-DA exhibited clear grouping between methanolic, and *n*-hexane extracts for fungal infections (Figure 7). The methanolic extract exhibited significant antifungal activity against *C. krusei*, the most sensitive yeast strain evaluated. The discriminant metabolites with the highest VIP scores comprised usnic acid,

vanillyl alcohol, and several flavonoid derivatives, recognized for their interference with ergosterol production and cell wall integrity. These processes are essential for antifungal effectiveness, especially in resistant *Candida* strains.



**Figure 7** PLS-DA of *Botryosphaeria rhodina* extracts against *Candida* species. (A) Methanol extract (M); (B) *n*-Hexane extract (H). Heatmaps show the relative abundance of discriminant metabolites across *Candida* strains: CA (*C. albicans*), CG (*C. glabrata*), and CK (*C. krusei*). VIP scores indicate the most influential metabolites contributing to class separation in the PLS-DA model.

In contrast, the *n*-hexane extract created a distinct cluster primarily composed of lipophilic molecules. The metabolites contributed minimally to class differentiation and exhibited insignificant connection with antifungal efficacy, indicating the restricted bioactivity of the non-polar fraction. The amalgamation of PLS-DA results with antimicrobial testing substantiates that polar metabolites in the methanol extract are the primary factors influencing both antibacterial and antifungal activity. This result aligns with the enhanced inhibition zones, reduced MIC/MBC/MFC values, and the significant chemical variety identified by LC-MS/MS in the methanolic fraction. Furthermore, the findings underscore the significant impact of solvent polarity on the metabolomic profile of endophytic fungus. Methanol extracts featured a chemically diversified array of bioactive polar molecules, whereas *n*-hexane extracts predominantly comprised lipophilic metabolites, which contributed minimally to antibacterial activity.

The PLS-DA model not only emphasizes the distinctions across solvent fractions but also provides prediction insights by associating specific metabolites

with bioactivity. Compounds exhibiting consistently elevated VIP scores—such as kojic acid, glycitein, fraxetin, deoxybrevianamide E, and usnic acid—represent potential subjects for additional research.

Their consistent contribution to class differentiation in bacterial and fungal datasets highlights their potential as bioactive candidates and indicates probable synergistic effects within the methanolic extract. These discoveries offer a logical foundation for focused isolation, structural determination, and mechanistic investigations, hence enhancing the bioprospecting significance of *B. rhodina*.

The utilization of PLS-DA enhances our investigation beyond mere descriptive metabolite profiling by establishing a statistically validated link between metabolite identification and biological function. The discovered clustering patterns and discriminant metabolites underscore possible synergistic interactions within the methanolic extract, necessitating additional separation and combination studies.

Collectively, these data affirm that *B. rhodina*, especially via its methanol-derived polar metabolites,

constitutes a prolific and chemically varied source of physiologically active secondary metabolites. Furthermore, the implementation of multivariate chemometric techniques like PLS-DA improves the dependability of natural product identification, hence augmenting the bioprospecting potential of endophytic fungus from mangrove habitats.

### Biosynthetic origins of identified metabolites

The metabolite diversity in *B. rhodina* can be elucidated by associating the identified chemical classes with fungal biosynthetic pathways and gene clusters (BGCs). In filamentous fungi, secondary metabolites are frequently encoded by clustered genes, including polyketide synthases (PKSs), nonribosomal peptide synthetases (NRPSs), hybrid PKS-NRPSs, and terpene synthases (TSs). Global transcription factors and environmental signals tightly control these clusters. They make sure that metabolite production is in sync with fungal growth and adaptation to the environment [43].

The study identified compounds derived from polyketides and phenolics, including kojic acid, fraxetin, and usnic acid, which are generally biosynthesized via PKS pathways. Recent metabolomic and genomic studies have shown that PKS-driven BGCs are still very important for the production of fungal aromatic polyketides. These BGCs are often turned on when nutrients are scarce or when oxidative stress is present. Usnic acid and structurally analogous dibenzofuran polyketides are associated with aromatic PKSs in conjunction with oxidative tailoring enzymes [44]. These pathways elucidate the enrichment of bioactive phenolic compounds in the methanol extract of *B. rhodina*.

The identification of flavonoid-like metabolites, such as glycitein, is notably intriguing due to their traditional correlation with plants. Nonetheless, recent investigations have demonstrated that endophytic fungi may possess PKS enzymes exhibiting chalcone synthase-like activity, potentially obtained via horizontal gene transfer or ecological adaptation to plant hosts [45,46]. This may elucidate the occurrence of isoflavonoid derivatives in mangrove-derived *B. rhodina*, indicating adaptation to its intricate ecological niche.

Indole alkaloids and diketopiperazines, such as deoxybrevianamide E, are typically synthesized by NRPSs or PKS-NRPS hybrids. Genome mining of fungal NRPS clusters has identified conserved pathways for indole alkaloid biosynthesis, especially within genera associated with *Aspergillus* and *Penicillium* [47]. The presence of deoxybrevianamide E in *B. rhodina* corroborates the hypothesis that similar NRPS-mediated pathways are operational in this species.

Moreover, depsidone-type metabolites, including botryorhodines previously identified in *B. rhodina*, constitute an additional category of PKS-derived compounds necessitating modifications mediated by methyltransferases and oxidases [48]. Even though we didn't find these in this study, the metabolic capacity shown by LC-MS/MS profiling suggests that *B. rhodina* may still have silent or conditionally expressed PKS clusters that can make depsidone.

The nonpolar fraction, which is rich in sterols and terpenoids (like clionasterol, squalene, and ergosta-4,7,22-trien-3-one), shows that the mevalonate pathway and fungal TS clusters are working. Recent research has underscored the ecological functions of fungal terpenoids in membrane stability, oxidative stress responses, and antifungal resistance [49]. Finding them here supports the idea that making terpenoids helps with both ecological fitness and the ability to fight off bacteria.

At the regulatory level, these biosynthetic processes are regulated by global transcriptional complexes like LaeA and the Velvet family proteins. Recent research validates that these regulators amalgamate environmental signals—such as light, salinity, and nutrient stress—to simultaneously activate or inhibit various BGCs [50]. The dynamic mangrove habitat of *B. rhodina*, characterized by salinity variations and microbial competition, may serve as a natural catalyst for the activation of cryptic or stress-responsive gene clusters, elucidating the diverse metabolome observed during solid-state fermentation.

In summary, the metabolite classes found in *B. rhodina* can be directly connected to PKS, NRPS, and TS gene clusters, as well as global regulatory networks. This molecular viewpoint emphasizes the genetic and ecological factors influencing metabolite diversity and underscores the importance of combining metabolomics

with genome mining to discover new antimicrobial agents.

### Limitations and future perspectives

The current work offers new insights into the antibacterial properties and metabolite composition of *B. rhodina*; nonetheless, numerous limitations require careful attention. Initially, metabolite identification predominantly depended on LC-HRMS and GC-MS data analyzed via spectrum libraries and *in silico* fragmentation predictions. This method facilitated the annotation of several metabolites; nevertheless, the lack of confirmatory tests, such as NMR or targeted HR-MS/MS, limits the confidence in structural assignments. This constraint is especially significant for over 25 unidentified metabolites detected in high relative abundance, several of which may signify novel chemical scaffolds. The exact function of these chemicals in antibacterial action remains uncertain without comprehensive structural elucidation.

Secondly, while the amalgamation of antimicrobial bioassays with PLS-DA analysis enhanced the correlation between metabolite composition and biological activity, the established relationships are qualitative rather than quantitative. The utilization of crude extracts hinders the assignment of activity to individual compounds and fails to consider possible synergistic or antagonistic interactions among metabolites. Bioassay-guided fractionation and purification are crucial for validating the roles of the discriminant metabolites found using chemometric modeling. Furthermore, synergy testing and combination assays should be utilized to elucidate the interactions among various metabolites that yield the observed antibacterial and antifungal activities.

The antimicrobial assessment was confined to a specific panel of bacterial and fungal pathogens. The results indicate substantial efficacy, especially against Gram-positive bacteria and *Candida* species; however, the effectiveness of *B. rhodina* metabolites against multidrug-resistant strains or clinically significant pathogens such as *Pseudomonas aeruginosa*, *Acinetobacter baumannii*, or *Cryptococcus neoformans* has yet to be investigated. Broadening the range of test organisms is essential for a more thorough evaluation of the therapeutic potential of these metabolites.

This study ultimately excluded genomic or transcriptome investigations, which could yield more profound insights into the biosynthetic gene clusters responsible for metabolite production. Integrating metabolomic data with genome mining and transcriptomics would enable the identification of cryptic or silent gene clusters and assist in the strategic modification of culture conditions to optimize the production of target metabolites. Subsequent research should examine co-culture or stress-induced methodologies to activate dormant biosynthetic pathways.

Collectively, these constraints underscore significant directions for future inquiry. Comprehensive structural elucidation, bioassay-guided isolation, synergy assessment, extensive antimicrobial screening, and integrative omics methodologies will be essential for progressing *B. rhodina* from a potential bioprospecting candidate to a confirmed source of new antimicrobial drugs.

This study highlights the significance of metabolites generated from endophytes for sustainable chemistry and pharmacy, in addition to their therapeutic consequences. The employment of endophytic fungus as sustainable microbial resources diminishes the necessity for massive plant harvesting, therefore preserving delicate ecosystems like mangroves. Our methodology—solid-state fermentation utilizing rice substrate—exhibits a low-energy, ecologically friendly, and scalable technique for metabolite production, in stark contrast to the resource-intensive submerged fermentation process. This study identifies bioactive molecules that may operate as environmentally friendly alternatives to synthetic antimicrobials, thereby alleviating the chemical and environmental loads linked to traditional medication manufacture. Our findings align with the tenets of green chemistry and directly endorse the United Nations Sustainable Development Goals, including SDG 3 (Good Health and Well-being) and SDG 12 (Responsible Consumption and Production). Consequently, the findings of this research transcend natural product discovery, illuminating a route toward a more sustainable bio-circular economy in pharmaceutical development.

## Conclusions

The compositional diversity and pharmacological potential of *B. rhodina* extracts underscore the importance of mangrove-derived endophytes as sustainable resources for antimicrobial drug discovery. The methanol extract, rich in polar secondary metabolites including kojic acid, usnic acid, fraxetin, and glycitein, exhibited significant antibacterial and mild antifungal properties, while chemometric tests indicated distinct correlations between metabolite profiles and bioactivity. The application of solid-state fermentation on rice substrate offers a low-energy, scalable, and ecologically friendly manufacturing method that adheres to the principles of green chemistry. This work promotes the use of endophytic fungi as sustainable microbiological sources, aiding the creation of environmentally friendly alternatives to synthetic antimicrobials, thereby advancing pharmaceutical innovation and biodiversity preservation.

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## Declaration of Generative AI in Scientific Writing

The author(s) employed the QuillBot tool solely to check the grammar and sentence structure while writing this manuscript. There was no AI tools used to create content, rephrase it, or analyze data. After using this tool, the author(s) carefully edited and reviewed the whole manuscript to make sure it was correct and clear. They take full responsibility for the content and conclusions in this publication.

## CRedit Author Statement

**Rudi Hendra:** Conceptualization; Writing - Original Draft; Funding. **Muhammad Rohim:** Investigation; formal analysis. **Yuli Haryani:** methodology; supervision. **Ari Satia Nugraha:** Writing - Original Draft; Writing - Review & Editing. **Hilwan**

**Yuda Terun:** Writing - Review & Editing. **Fauzan Zein Muttaqin:** Validation; software; visualization.

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